

=> d his

(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1 44 S ACYLAminoACID/IA
L2 5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?) /IA
L3 0 S L1 AND L2
L4 113543 S DETERGENT
L5 3 S L1 AND L4
L6 7262 S PHOSPHOROUS (2W) ACID
L7 1836 S HYPOPHOSPHOROUS (2W) ACID
L8 7262 S (L6 OR L7 (6W) ODOR)
L9 2 S L8 (5W) MEMBRANE
L10 73 S L8 AND L4
L11 0 S L10 AND L1
L12 0 S L10 (5W) ACYLAminoACID
L13 7262 S (L6 OR L7 (2W) ODOR)
L14 7273 S (L6 OR L7 AND ODOR)
L15 7263 DUP REM L14 (10 DUPLICATES REMOVED)
L16 113867 S GLUTAMIC
L17 7263 S L15
L18 23 S L15 AND L16
L19 40 S FATTYACID
L20 0 S L19 AND HALIDE
L21 676258 S REMOVAL
L22 7263 S L15
L23 311 S L21 AND L15
L24 2 S L23 (4W) L16
L25 78807 S ODOR
L26 7066 S L25 AND L21
L27 4 S L26 AND L7

FILE 'USPATFULL' ENTERED AT 17:30:45 ON 21 AUG 2007

L28 11772 S DEODORANT
L29 144537 S PHOSPHORUS
L30 61092 S ODOR
L31 956265 S REMOVAL
L32 46 S (L29 (4W) L31 (4W) L29)
L33 0 S L28 AND L32
L34 83 S ACYLAminoACID
L35 4 S L34 AND L29
L36 0 S L35 AND L30
L37 1 S L35 AND L31

FILE 'CAPLUS' ENTERED AT 17:38:19 ON 21 AUG 2007

L38 8 S L26 AND L6

FILE 'STNGUIDE' ENTERED AT 17:44:31 ON 21 AUG 2007

=> log off

ALL L# QUERIES AND ANSWER SETS ARE DELETED AT LOGOFF

LOGOFF? (Y)/N/HOLD:Y

STN INTERNATIONAL LOGOFF AT 17:47:09 ON 21 AUG 2007

Connecting via Winsock to STN

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LOGINID:SSPTAYKC1621

PASSWORD :

TERMINAL (ENTER 1, 2, 3, OR ?):2

NEWS 1 Web Page for STN Seminar Schedule - N. America
NEWS 2 MAY 01 New CAS web site launched
NEWS 3 MAY 08 CA/CAPLUS Indian patent publication number format defined
NEWS 4 MAY 14 RDISCLOSURE on STN Easy enhanced with new search and display fields
NEWS 5 MAY 21 BIOSIS reloaded and enhanced with archival data
NEWS 6 MAY 21 TOX CENTER enhanced with BIOSIS reload
NEWS 7 MAY 21 CA/CAPLUS enhanced with additional kind codes for German patents
NEWS 8 MAY 22 CA/CAPLUS enhanced with IPC reclassification in Japanese patents
NEWS 9 JUN 27 CA/CAPLUS enhanced with pre-1967 CAS Registry Numbers
NEWS 10 JUN 29 STN Viewer now available
NEWS 11 JUN 29 STN Express, Version 8.2, now available
NEWS 12 JUL 02 LEMBASE coverage updated
NEWS 13 JUL 02 LMEDLINE coverage updated
NEWS 14 JUL 02 SCISEARCH enhanced with complete author names
NEWS 15 JUL 02 CHEMCATS accession numbers revised
NEWS 16 JUL 02 CA/CAPLUS enhanced with utility model patents from China
NEWS 17 JUL 16 CAPLUS enhanced with French and German abstracts
NEWS 18 JUL 18 CA/CAPLUS patent coverage enhanced
NEWS 19 JUL 26 USPATFULL/USPAT2 enhanced with IPC reclassification
NEWS 20 JUL 30 USGENE now available on STN
NEWS 21 AUG 06 CAS REGISTRY enhanced with new experimental property tags
NEWS 22 AUG 06 BEILSTEIN updated with new compounds
NEWS 23 AUG 06 FSTA enhanced with new thesaurus edition
NEWS 24 AUG 13 CA/CAPLUS enhanced with additional kind codes for granted patents
NEWS 25 AUG 20 CA/CAPLUS enhanced with CAS indexing in pre-1907 records

NEWS EXPRESS 29 JUNE 2007: CURRENT WINDOWS VERSION IS V8.2,
CURRENT MACINTOSH VERSION IS V6.0c(ENG) AND V6.0Jc(JP),
AND CURRENT DISCOVER FILE IS DATED 05 JULY 2007.

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NEWS LOGIN Welcome Banner and News Items
NEWS IPC8 For general information regarding STN implementation of IPC 8

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FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007
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FILE COVERS 1907 - 21 Aug 2007 VOL 147 ISS 9
FILE LAST UPDATED: 20 Aug 2007 (20070820/ED)

Effective October 17, 2005, revised CAS Information Use Policies apply.
They are available for your review at:

<http://www.cas.org/infopolicy.html>

=> s acylaminoacid/IA
 32 ACYLAminoACID/IA
 12 ACYLAminoACIDS/IA
L1 44 ACYLAminoACID/IA
 ((ACYLAminoACID OR ACYLAminoACIDS)/IA)

```
=> s (fragrance (2w) cosmetic (2w) toiletr?)/IA  
      12559 FRAGRANCE/IA  
      3233 FRAGRANCES/IA  
      14191 FRAGRANCE/IA  
              ((FRAGRANCE OR FRAGRANCES)/IA)  
60921 COSMETIC/IA  
66152 COSMETICS/IA  
84925 COSMETIC/IA  
              ((COSMETIC OR COSMETICS)/IA)  
607 TOILETR?/IA  
      5 (FRAGRANCE (2W) COSMETIC (2W) TOILETR?)/IA
```

=> s l1 and l2
L3 0 L1 AND L2

=> d 12 1-5 IBIB

L2 ANSWER 1 OF 5 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2007:17115 CAPLUS
 DOCUMENT NUMBER: 146:102730
 TITLE: Substituted cyclopentenone derivatives and their use
 in perfume compositions
 INVENTOR(S): Levorse, Anthony T.; Weiss, Richard A.; Newirth, Brett
 D.
 PATENT ASSIGNEE(S): USA
 SOURCE: U.S. Pat. Appl. Publ., 5pp., Cont.-in-part of U.S.
 Ser. No. 173,539.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2007004608	A1	20070104	US 2006-419874	20060523
US 7141699	B1	20061128	US 2005-173539	20050701
PRIORITY APPLN. INFO.:			US 2005-173539	A2 20050701

L2 ANSWER 2 OF 5 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2006:1243068 CAPLUS
 DOCUMENT NUMBER: 146:7632
 TITLE: Preparation of substituted cyclopentenone derivatives
 and their use in perfume and cleaning compositions
 INVENTOR(S): Levorse, Anthony T., Jr.; Weiss, Richard A.; Newirth,
 Brett D.
 PATENT ASSIGNEE(S): International Flavors & Fragrances Inc., USA
 SOURCE: U.S., 4pp.
 CODEN: USXXAM
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 2
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 7141699	B1	20061128	US 2005-173539	20050701
US 2007004608	A1	20070104	US 2006-419874	20060523
PRIORITY APPLN. INFO.:			US 2005-173539	A2 20050701
OTHER SOURCE(S):	CASREACT 146:7632; MARPAT 146:7632			
REFERENCE COUNT:	3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT			

L2 ANSWER 3 OF 5 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2006:1091384 CAPLUS
 DOCUMENT NUMBER: 145:418931
 TITLE: 3-Methyl-3-oxetanemethanol derivatives and their
 preparation and use in perfume compositions
 INVENTOR(S): Levorse, Anthony T.; Mertz, Gary
 PATENT ASSIGNEE(S): International Flavors & Fragrances, Inc., USA
 SOURCE: Eur. Pat. Appl., 14pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1712551	A1	20061018	EP 2006-251697	20060329
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, PL, SK, BA, HR, IS, YU				
US 2006234882	A1	20061019	US 2005-105625	20050414
PRIORITY APPLN. INFO.: US 2005-105625 A 20050414				
OTHER SOURCE(S): MARPAT 145:418931				
REFERENCE COUNT: 6 THERE ARE 6 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT				

L2 ANSWER 4 OF 5 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2002:183744 CAPLUS
 DOCUMENT NUMBER: 136:236693
 TITLE: Enhancing diffusivity and long-lasting property of
fragrances
 INVENTOR(S): Ishida, Kenya; Matsuda, Hiroyuki; Iwai, Hisao;
Hagiwara, Toshimitsu; Nishida, Takashi
 PATENT ASSIGNEE(S): Takasago International Corporation, Japan
 SOURCE: Eur. Pat. Appl., 29 pp.
 CODEN: EPXXDW
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1186289	A2	20020313	EP 2001-402340	20010911
EP 1186289	A3	20030423		
EP 1186289	B1	20070207		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
JP 2002088391	A	20020327	JP 2000-275928	20000912
AU 200169989	A	20020314	AU 2001-69989	20010911
AU 780080	B2	20050224		
CA 2357251	A1	20020312	CA 2001-2357251	20010912
CN 1343487	A	20020410	CN 2001-133066	20010912
US 2002054893	A1	20020509	US 2001-949718	20010912
JP 2000-275928 A 20000912				
PRIORITY APPLN. INFO.: MARPAT 136:236693				

L2 ANSWER 5 OF 5 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1984:635357 CAPLUS
 DOCUMENT NUMBER: 101:235357
 TITLE: Dangers in perfume toxicity: some indications for the
mass production industries
 AUTHOR(S): Savage, Paul Kevin
 CORPORATE SOURCE: Bush Boake Allen Ltd., London, UK
 SOURCE: Aerosol Report (1984), 23(6), 303-4, 306-7, 309-11
 DOCUMENT TYPE: Journal; General Review
 LANGUAGE: English/German

=> s detergent

L4 82983 DETERGENT
 72495 DETERGENTS
 113543 DETERGENT
 (DETERGENT OR DETERGENTS)

=> s 11 and 14
 L5 3 L1 AND L4

=> d 15 1 IBIB

L5 ANSWER 1 OF 3 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2003:525600 CAPLUS
 DOCUMENT NUMBER: 139:87029
 TITLE: Cleaner compositions containing urea and ethanol amine salts
 INVENTOR(S): Saito, Yoshinobu; Saito, Junko; Nishina, Tetsuo
 PATENT ASSIGNEE(S): Shiseido Honey Cake Factory K. K., Japan; Shiseido Honeycake Industries Co., Ltd.
 SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2003193092	A	20030709	JP 2001-397266	20011227
JP 3547728	B2	20040728		
PRIORITY APPLN. INFO.:			JP 2001-397266	20011227

=> d 15 2 IBIB

L5 ANSWER 2 OF 3 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2001:366166 CAPLUS
 DOCUMENT NUMBER: 134:368648
 TITLE: Solid cleaning compositions containing N-acylamino acids
 INVENTOR(S): Komatsu, Atsushi; Suzuki, Takayuki
 PATENT ASSIGNEE(S): Kawaken Fine Chemicals Co., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 10 pp.
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2001139985	A	20010522	JP 1999-327800	19991118
PRIORITY APPLN. INFO.:			JP 1999-327800	19991118
OTHER SOURCE(S):			MARPAT 134:368648	

=> d 15 3 IBIB

L5 ANSWER 3 OF 3 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1997:303066 CAPLUS

DOCUMENT NUMBER: 126:279348
 TITLE: Liquid detergent compositions having proper consistency
 INVENTOR(S): Umemoto, Isao; Kajiwara, Yasushi
 PATENT ASSIGNEE(S): Kao Corp, Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 7 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 09067591	A	19970311	JP 1995-248321	19950904
US 5767059	A	19980616	US 1996-703535	19960827
CN 1149619	A	19970514	CN 1996-113234	19960904
CN 1083882	B	20020501		
PRIORITY APPLN. INFO.:			JP 1995-248321	A 19950904
OTHER SOURCE(S):	MARPAT	126:279348		

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(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1 44 S ACYLAAMINOACID/IA
 L2 5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?)/IA
 L3 0 S L1 AND L2
 L4 113543 S DETERGENT
 L5 3 S L1 AND L4

=> s phosphorous (2w) acid
 20699 PHOSPHOROUS
 4422696 ACID
 1587666 ACIDS
 4924341 ACID
 (ACID OR ACIDS)
 L6 7262 PHOSPHOROUS (2W) ACID

=> s hypophosphorous (2w) acid
 1863 HYPOPHOSPHOROUS
 4422696 ACID
 1587666 ACIDS
 4924341 ACID
 (ACID OR ACIDS)
 L7 1836 HYPOPHOSPHOROUS (2W) ACID

=> s (16 or 17 (6w) odor)
 73557 ODOR
 12714 ODORS
 78807 ODOR
 (ODOR OR ODORS)
 L8 0 L7 (6W) ODOR
 7262 (L6 OR L7 (6W) ODOR)

=> s 18 (5w) membrane
 758488 MEMBRANE

329064 MEMBRANES
 848453 MEMBRANE
 (MEMBRANE OR MEMBRANES)
 L9 2 L8 (5W) MEMBRANE

=> d 19 1-2 IBIB abs

L9 ANSWER 1 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2005:1220136 CAPLUS
 DOCUMENT NUMBER: 143:463153
 TITLE: Proton-conductive membranes, catalyst electrode-proton conductor assemblies, and fuel cells
 INVENTOR(S): Matsuo, Kazumine; Kin, Shinichiro; Sano, Hiroki; Omichi, Takahiro
 PATENT ASSIGNEE(S): Teijin Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 21 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	-----	-----	-----	-----
JP 2005320472	A	20051117	JP 2004-140958	20040511
PRIORITY APPLN. INFO.:			JP 2004-140958	20040511

AB The H+-conductive membranes are obtained by hydrolysis and condensation of amino-containing Si alkoxides, amino-free Si alkoxides, metal alkoxides containing

Ti, Al, and/or Zr, and phosphate compds. or phosphite compds. to prepare a 1st solution containing metal oxide derivs., adding the 1st solution to a solution

containing H+-conductive organic polymers having T (temperature where main dispersion

of mol. chains is observed by dynamic viscoelastic measurement)

60-270° to prepare a 2nd solution containing the H+-conductive organic polymers and metal oxide derivs., and casting the 2nd solution, and show ≤90% decrease in storage modulus at T(°) compared to that at 30°.

The catalyst electrode-proton conductor assemblies have catalyst electrodes comprising metals supported on elec. conductive particulate carriers on both sides of the H+-conductive membranes. The H+-conductive membranes are MeOH-insol., show good film-forming properties and H+ conductivity,

suppress crossover of MeOH, and are useful for direct-methanol polymer electrolyte fuel cells.

L9 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2005:368892 CAPLUS
 DOCUMENT NUMBER: 142:384650
 TITLE: Oil/gas separation membrane, its use in gas sensor, and process for producing the same
 INVENTOR(S): Qin, Ren Yan
 PATENT ASSIGNEE(S): Peop. Rep. China
 SOURCE: U.S. Pat. Appl. Publ., 14 pp.
 CODEN: USXXCO
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 2005086998	A1	20050428	US 2004-971443	20041021
CN 1608719	A	20050427	CN 2003-10111953	20031024
CA 2485808	A1	20050424	CA 2004-2485808	20041022
PRIORITY APPLN. INFO.:			CN 2003-10111953	A 20031024

AB An oil gas separation membrane combines a gas permeable yet oil and temperature resistant bulk polymer membrane such as poly(tetrafluoroethylene) and poly(tetrafluoroethylene-co-hexafluoropropylene); a porous metal support such as sintered metal frit disk made with stainless steel, bronze or nickel; and an highly gas permeable adhesive that bonds firmly the bulk polymer membrane and the metal frit surface together. The adhesive is either a homogeneous polymer that has desirable gas permeability, or a coalescent porous polymer particulates network. A gas sensor employing the oil gas separation membrane for detecting and monitoring fault gases of oil filled elec. equipment requires no mech. wearing or moving part such as pump and valve and the gas sensor is operated normally under various temperature and pressure conditions.

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(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1	44 S ACYLAminoACID/IA
L2	5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?)/IA
L3	0 S L1 AND L2
L4	113543 S DETERGENT
L5	3 S L1 AND L4
L6	7262 S PHOSPHOROUS (2W) ACID
L7	1836 S HYPOPHOSPHOROUS (2W) ACID
L8	7262 S (L6 OR L7 (6W) ODOR)
L9	2 S L8 (5W) MEMBRANE

=> s 18 and 14

L10 73 L8 AND L4

=> s 110 and 11

L11 0 L10 AND L1

=> s 110 (5w) acylaminoacid

PROXIMITY OPERATOR LEVEL NOT CONSISTENT WITH
FIELD CODE - 'AND' OPERATOR ASSUMED 'L10 (5W) ACYLAmino'

32 ACYLAminoACID
12 ACYLAminoACIDS
44 ACYLAminoACID
(ACYLAminoACID OR ACYLAminoACIDS)

L12 0 L10 (5W) ACYLAminoACID

=> s (16 or 17 (2w) odor)/ia

QUALIFICATION NOT VALID FOR L6

Field code qualifications can only be applied to text terms.

=> s (16 or 17 (2w) odor)

73557 ODOR

```

12714 ODORS
78807 ODOR
    (ODOR OR ODORS)
    0 L7 (2W) ODOR
L13     7262 (L6 OR L7 (2W) ODOR)

=> s (l6 or L7 and odor)
73557 ODOR
12714 ODORS
78807 ODOR
    (ODOR OR ODORS)
L14     7273 (L6 OR L7 AND ODOR)

=> dup rem l14
PROCESSING IS APPROXIMATELY 22% COMPLETE FOR L14
PROCESSING IS APPROXIMATELY 45% COMPLETE FOR L14
PROCESSING IS APPROXIMATELY 71% COMPLETE FOR L14
PROCESSING COMPLETED FOR L14
L15     7263 DUP REM L14 (10 DUPLICATES REMOVED)

=> s glutamic
    113867 GLUTAMIC
    1 GLUTAMICS
L16     113867 GLUTAMIC
        (GLUTAMIC OR GLUTAMICS)

=> s l15 and l16
L17     7263 S L15
L18     23 L17 AND L16

=> d l18 1 - 6 IBIB
'-' IS NOT A VALID FORMAT FOR FILE 'CAPLUS'

The following are valid formats:

ABS ----- GI and AB
ALL ----- BIB, AB, IND, RE
APPS ----- AI, PRAI
BIB ----- AN, plus Bibliographic Data and PI table (default)
CAN ----- List of CA abstract numbers without answer numbers
CBIB ----- AN, plus Compressed Bibliographic Data
CLASS ----- IPC, NCL, ECLA, FTERM
DALL ----- ALL, delimited (end of each field identified)
DMAX ----- MAX, delimited for post-processing
FAM ----- AN, PI and PRAI in table, plus Patent Family data
FBIB ----- AN, BIB, plus Patent FAM
IND ----- Indexing data
IPC ----- International Patent Classifications
MAX ----- ALL, plus Patent FAM, RE
PATTS ----- PI, SO
SAM ----- CC, SX, TI, ST, IT
SCAN ----- CC, SX, TI, ST, IT (random display, no answer numbers;
            SCAN must be entered on the same line as the DISPLAY,
            e.g., D SCAN or DISPLAY SCAN)
STD ----- BIB, CLASS

IABS ----- ABS, indented with text labels
IALL ----- ALL, indented with text labels
IBIB ----- BIB, indented with text labels

```

IMAX ----- MAX, indented with text labels
ISTD ----- STD, indented with text labels

OBIB ----- AN, plus Bibliographic Data (original)
OIBIB ----- OBIB, indented with text labels

SBIB ----- BIB, no citations
SIBIB ----- IBIB, no citations

HIT ----- Fields containing hit terms
HITIND ----- IC, ICA, ICI, NCL, CC and index field (ST and IT)
containing hit terms
HITRN ----- HIT RN and its text modification
HITSTR ----- HIT RN, its text modification, its CA index name, and
its structure diagram
HITSEQ ----- HIT RN, its text modification, its CA index name, its
structure diagram, plus NTE and SEQ fields
FHITSTR ----- First HIT RN, its text modification, its CA index name, and
its structure diagram
FHITSEQ ----- First HIT RN, its text modification, its CA index name, its
structure diagram, plus NTE and SEQ fields
KWIC ----- Hit term plus 20 words on either side
OCC ----- Number of occurrence of hit term and field in which it occurs

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ENTER DISPLAY FORMAT (BIB):IBIB

L18 ANSWER 1 OF 23 CAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2007:778682 CAPLUS
TITLE: Microwave-assisted synthesis of modified polyaspartic acid in solvent
AUTHOR(S): Zhang, Yuling; Huang, Junli; Cheng, Zhihui; Yang, Shilin
CORPORATE SOURCE: School of Municipal and Environmental Engineering, Harbin Institute of Technology, Harbin, 150090, Peop. Rep. China
SOURCE: Chinese Journal of Chemical Engineering (2007), 15(3), 458-462
CODEN: CJCEEB; ISSN: 1004-9541
PUBLISHER: Chemical Industry Press
DOCUMENT TYPE: Journal
LANGUAGE: English

L18 ANSWER 6 OF 23 CAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2005:259987 CAPLUS
DOCUMENT NUMBER: 142:321282
TITLE: Setting and hardening accelerator compositions containing α -amino acid for improved storage stability and performance in cementitious compositions
INVENTOR(S): Weibel, Martin

PATENT ASSIGNEE(S) : Construction Research & Technology GmbH, Germany
 SOURCE: PCT Int. Appl., 17 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2005026072	A1	20050324	WO 2004-EP9255	20040818
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
AU 2004272177	A1	20050324	AU 2004-272177	20040818
CA 2538313	A1	20050324	CA 2004-2538313	20040818
EP 1663904	A1	20060607	EP 2004-764241	20040818
EP 1663904	B1	20070214		
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK				
CN 1849275	A	20061018	CN 2004-80026337	20040818
JP 2007505024	T	20070308	JP 2006-525672	20040818
AT 353859	T	20070315	AT 2004-764241	20040818
NO 2006001590	A	20060612	NO 2006-1590	20060407
US 2007144405	A1	20070628	US 2007-569299	20070130
PRIORITY APPLN. INFO.:			GB 2003-21331	A 20030912
			WO 2004-EP9255	W 20040818

REFERENCE COUNT: 4 THERE ARE 4 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

```
=> s fattyacid
      24 FATTYACID
      16 FATTYACIDS
L19      40 FATTYACID
          (FATTYACID OR FATTYACIDS)
```

```
=> s l19 and halide
      154668 HALIDE
      130629 HALIDES
      225786 HALIDE
          (HALIDE OR HALIDES)
L20      0 L19 AND HALIDE
```

```
=> s removal
      674921 REMOVAL
      5536 REMOVALS
L21      676258 REMOVAL
          (REMOVAL OR REMOVALS)
```

```
=> d his
```

(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1 44 S ACYLAminoACID/IA
L2 5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?) /IA
L3 0 S L1 AND L2
L4 113543 S DETERGENT
L5 3 S L1 AND L4
L6 7262 S PHOSPHOROUS (2W) ACID
L7 1836 S HYPOPHOSPHOROUS (2W) ACID
L8 7262 S (L6 OR L7 (6W) ODOR)
L9 2 S L8 (5W) MEMBRANE
L10 73 S L8 AND L4
L11 0 S L10 AND L1
L12 0 S L10 (5W) ACYLAminoACID
L13 7262 S (L6 OR L7 (2W) ODOR)
L14 7273 S (L6 OR L7 AND ODOR)
L15 7263 DUP REM L14 (10 DUPLICATES REMOVED)
L16 113867 S GLUTAMIC
L17 7263 S L15
L18 23 S L15 AND L16
L19 40 S FATTYACID
L20 0 S L19 AND HALIDE
L21 676258 S REMOVAL

=> s 121 and 115

L22 7263 S L15
L23 311 L21 AND L22

=> s 123 (4w) 116

PROXIMITY OPERATOR LEVEL NOT CONSISTENT WITH
FIELD CODE - 'AND' OPERATOR ASSUMED 'L23 (4W) L16'
L24 2 L23 (4W) L16

=> d 124 1-2 IBIB abs

L24 ANSWER 1 OF 2 CPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 2001:582013 CPLUS
DOCUMENT NUMBER: 135:154426
TITLE: Hard surface cleaning and disinfecting composition
INVENTOR(S): Urban, Virginia Lee
PATENT ASSIGNEE(S): Reckitt Benckiser Inc., USA; Reckitt Benckiser (UK)
 Limited
SOURCE: PCT Int. Appl., 34 pp.
 CODEN: PIXXD2
DOCUMENT TYPE: Patent
LANGUAGE: English
FAMILY ACC. NUM. COUNT: 1
PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
WO 2001057174	A1	20010809	WO 2001-GB384	20010131
W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CR, CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN,				

YU, ZA, ZW
 RW: GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY,
 DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR, BF,
 BJ, CF, CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
 CA 2396742 A1 20010809 CA 2001-2396742 20010131
 GB 2360786 A 20011003 GB 2001-2426 20010131
 GB 2360786 B 20020424
 EP 1252283 A1 20021030 EP 2001-904066 20010131
 EP 1252283 B1 20070711
 R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
 IE, SI, LT, LV, FI, RO, MK, CY, AL, TR
 US 2002187918 A1 20021212 US 2002-210562 20020731
 US 6936579 B2 20050830
 PRIORITY APPLN. INFO.: GB 2000-2229 A 20000201
 WO 2001-GB384 W 20010131

AB An aqueous hard surface cleaning and disinfecting composition comprises: a first

acid sequestrant constituent which comprises citric acid and a second acid sequestrant which is selected from citric acid, cresylic acid, dodecylbenzene sulfonic acid, phosphoric acid, salicylic acid, sorbic acid, sulfamic acid, acetic acid, benzoic acid, boric acid, capric acid, caproic acid, cyanuric acid, dihydroacetic acid, dimethylsulfamic acid, propionic acid, polyacrylic acid, 2-Et hexanoic acid, formic acid, fumaric acid, 1-glutamic acid, iso-Pr sulfamic acid, naphthenic acid, oxalic acid, phosphorus acid, valeric acid, benzene sulfonic acid, xylene sulfonic acid, sulfonic acids, maleic acid, acetic acid, adipic acid, lactic acid, butyric acid, gluconic acid, malic acid, tartaric acid, and glycolic acid; a mixture of hydrophobic and hydrophilic solvents wherein the hydrophobic solvent exhibits a solubility in water of from 0.0 - 20.0 mL/100mL of water, and which comprises 51-99% of the mixture of solvents, and wherein the hydrophilic solvent comprises 1-49% of the mixture of solvents; 0.001-1% by weight of a single constituent which exhibits both anionic surfactant and hydrotrope properties; 0-20% of one or more optional constituents; the balance to 100% by weight, water; wherein the composition has pH ≤7.0. The compns. provide good removal of soap scum stains, and further feature low levels of irritability to the user. Although acidic, the improved hard surface cleaning compns. feature low irritability to the eyes and skin of consumers. The compns. also provide disinfecting effects. Processes for the production of the compns., as well as methods for their use are also described.

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 2 OF 2 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1997:397379 CAPLUS
 DOCUMENT NUMBER: 127:19997
 TITLE: Germicidal acidic hard surface cleaning compositions
 for removal of soap scum stains
 INVENTOR(S): Crisanti, Michael
 PATENT ASSIGNEE(S): Reckitt & Colman Inc., USA
 SOURCE: PCT Int. Appl., 28 pp.
 CODEN: PIXXD2
 DOCUMENT TYPE: Patent
 LANGUAGE: English
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
-----	-----	-----	-----	-----

WO 9715649	A1	19970501	WO 1996-US15463	19960927
W: AL, AM, AU, BB, BG, BR, CA, CN, CZ, EE, FI, GE, HU, IL, IS, JP, KG, KP, KR, LK, LR, LT, LV, MD, MG, MK, MN, MX, NO, NZ, PL, RO, SD, SG, SI, SK, TR, TT, UA, UG, UZ, VN, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
RW: KE, LS, MW, SD, SZ, UG, AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF, CG, CI, CM, GA, GN, ML, MR, NE, SN, TD, TG				
GB 2306499	A	19970507	GB 1995-21829	19951025
GB 2306500	A	19970507	GB 1996-12645	19960617
GB 2306500	B	20000223		
US 6221823	B1	20010424	US 1996-709759	19960909
CA 2235484	A1	19970501	CA 1996-2235484	19960927
AU 9673741	A	19970515	AU 1996-73741	19960927
AU 718194	B2	20000406		
CN 1202925	A	19981223	CN 1996-198430	19960927
CN 1088748	B	20020807		
EP 904343	A1	19990331	EP 1996-935987	19960927
EP 904343	B1	20040211		
R: BE, DE, ES, FR, GB, IT, NL				
BR 9611215	A	19990601	BR 1996-11215	19960927
ES 2211989	T3	20040716	ES 1996-935987	19960927
ZA 9608888	A	19970805	ZA 1996-8888	19961023
PRIORITY APPLN. INFO.:			GB 1995-21829	A 19951025
			GB 1996-12645	A 19960617
			WO 1996-US15463	W 19960927

AB The compns., having minimal irritability to skin or eyes and pH ≤5.0, comprise 0.1-10% an acid sequestrant; 0.1-10% mixture of hydrophobic and hydrophilic solvents; 1-8% surfactant and/or hydrotrope; 0-20% ≥1 optional constituents; the balance 100% H₂O. A cleaner containing Polytergent SL 62 1.00, Rhodapon LCP 3.00, Polytergent 2A1 3.00, Dowanol PnP 0.90, Dowanol PnB 3.90, citric acid 2.50, glycolic acid 3.57, fragrance 0.20, and water 81.93 had 63% cleaning efficiency (calculated from reflectance values); vs. 2.1% for a cleaner without glycolic acid or citric acid.

=> d his

(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1	44 S ACYLAminoACID/IA
L2	5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?)/IA
L3	0 S L1 AND L2
L4	113543 S DETERGENT
L5	3 S L1 AND L4
L6	7262 S PHOSPHOROUS (2W) ACID
L7	1836 S HYPOPHOSPHOROUS (2W) ACID
L8	7262 S (L6 OR L7 (6W) ODOR)
L9	2 S L8 (5W) MEMBRANE
L10	73 S L8 AND L4
L11	0 S L10 AND L1
L12	0 S L10 (5W) ACYLAminoACID
L13	7262 S (L6 OR L7 (2W) ODOR)
L14	7273 S (L6 OR L7 AND ODOR)
L15	7263 DUP REM L14 (10 DUPLICATES REMOVED)
L16	113867 S GLUTAMIC
L17	7263 S L15

L18 23 S L15 AND L16
 L19 40 S FATTYACID
 L20 0 S L19 AND HALIDE
 L21 676258 S REMOVAL
 L22 7263 S L15
 L23 311 S L21 AND L15
 L24 2 S L23 (4W) L16

=> s odor
 73557 ODOR
 12714 ODORS
 L25 78807 ODOR
 (ODOR OR ODORS)

=> s 125 and 121
 L26 7066 L25 AND L21

=> s 126 and 17
 L27 4 L26 AND L7

=> d 127 1-4 IBIB abs

L27 ANSWER 1 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 2007:144504 CAPLUS
 DOCUMENT NUMBER: 146:207932
 TITLE: Fiber products with washfast deodorant function,
 comprising cotton fiber products crosslinked with
 polycarboxylic acids and manufacture thereof
 INVENTOR(S): Akima, Mitsuru; Kuroda, Toshinori
 PATENT ASSIGNEE(S): Akkus K. K., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 2007031853	A	20070208	JP 2005-213522	20050722
PRIORITY APPLN. INFO.:			JP 2005-213522	20050722

AB The fiber products (A1) comprise fiber products mainly containing cotton and crosslinked with polycarboxylic acids, or the fiber products comprise above A1 fiber products holding metal salts. The fiber products are prepared by the steps comprising the steps of (a) immersing a fiber product mainly containing cotton in a mixture containing a polycarboxylic acid, a hypophosphorous acid salt, and a metal or metal salt from Zn, Fe, Cu, Al, Co, and Ag, and (b) drying the product to simultaneously crosslink the fiber product with the polycarboxylic acid and cause the fiber product holding the metal or metal salt. The deodorant fiber products are useful for beddings for old people. A cotton fiber product was padded with a liquid containing 50:50:50 (weight ratio) mixture of citric acid, butanetetracarboxylic acid, and sodium phosphinate, an acrylic polymer, and 50 parts copper acetate, dried, heat-treated 3 min at 160°, to give a fabric showing NH₃ odor removal degree 99% and H₂S odor removal degree 85%, and showing NH₃ odor removal degree 99 and 96%, resp., after 10 and 20 washings.

L27 ANSWER 2 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1990:617267 CAPLUS
 DOCUMENT NUMBER: 113:217267
 TITLE: Deodorant and method for removing odors from tank trucks for collection of night soil by the vacuum method
 INVENTOR(S): Yoshida, Norikazu; Matsumoto, Susumu
 PATENT ASSIGNEE(S): Nippon Kayaku Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 02160019	A	19900620	JP 1988-312845	19881213
PRIORITY APPLN. INFO.:			JP 1988-312845	19881213

AB The deodorant comprising ≥ 1 of phosphoric acid, phosphorous acid, hypophosphorous acid, and peroxide. Thus, 50% phosphorous acid 200 g, activated C 100 g, and Na CM-cellulose 100 g were kneaded with a little water, extruded into 5 mm pellets, and dried at 110° to obtain a deodorant. A model exhaust gas (simulated night soil vacuum pump waste gas saturated with H₂O at 40°) containing H₂S 6000, CH₃SH 100, and NH₃ 300 ppm was passed through a column packed with 1200 mL of the deodorant at 42 L/min (contact time 1.7 s). The time required for 10% breakthrough for H₂S, CH₃SH, and NH₃ was 0.3, 20.4, and 39.1 h, resp., compared with 0.2, 7.4, and 8.1 h, resp., for a granular activated C as a control.

L27 ANSWER 3 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1990:164276 CAPLUS
 DOCUMENT NUMBER: 112:164276
 TITLE: Deodorant compositions capable of removing odors of basic materials and mercaptans
 INVENTOR(S): Miki, Yoshiaki; Ueda, Tsunehisa; Natsume, Yoshio
 PATENT ASSIGNEE(S): Nippon Zeon Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 3 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01124459	A	19890517	JP 1987-281371	19871107
JP 2549876	B2	19961030		

PRIORITY APPLN. INFO.:

AB The compns. comprise a phosphite or hypophosphite, a Cu compound, and a solvent. Mixts. of H₃PO₃, H₃PO₂, (EtO)₂PO, Na hypophosphite, Pb hypophosphite, CuSO₄, CuCl₂, NaCu chlorophyllin, and Cu oleate in H₂O and DOP were used for removal of MeSH in examples.

L27 ANSWER 4 OF 4 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1922:277 CAPLUS
 DOCUMENT NUMBER: 16:277

ORIGINAL REFERENCE NO.: 16:54b-h
 TITLE: Action of phosphine on formaldehyde
 AUTHOR(S): Hoffman, Alfred
 SOURCE: Journal of the American Chemical Society (1921), 43,
 1684-8
 CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal
 LANGUAGE: Unavailable

AB The PH₃ was prepared by igniting a paper containing a mixture of 2 parts Al powder and 1 part red P and treating the resulting spongy AlP with H₂O; the gas so obtained does not ignite spontaneously at room temperature. When it is passed through 90 cc. of 40% HCHO and 40 cc. HCl (d. 1.2) at 80° and the liquid is afterwards evaporated to dryness on the H₂O bath there is obtained 50 g. of a substance having the properties of a tetrahydroxymethylenephosphonium chloride (A), (HOCH₂)₄PCl, flat low-melting needles with solvent of crystallization from AcOH, m. 151° after heating in dry air at 100°, very deliquescent, forms oily products with AcCl and Ac₂O, can be boiled in H₂O without decomposition, is not affected by dilute acids but is extremely sensitive towards alkalies and even neutral carbonates; with NaOH, even in the cold, H is vigorously evolved and if the reaction is completed by warming almost exactly 4 atoms H are split off and on acidifying and distilling 0.5 mol. HCO₂H comes over. The residue of the mol. containing the P is more easily isolated when the decomposition is effected with Ba(OH)₂; after removal of the Ba and evaporation there remains a thick non-crystallizable sirup which, according to the yield and the P content, apparently has the empirical formula C₃H₉O₅P and is possibly (HOCH₂)₃PO₂; perhaps it is formed according to the equation (HOCH₂)₄PCl + NaOH + H₂O = (HOCH₂)₃PO₂ + NaCl + 2H₂ + HCHO, the HCHO then reacting with the NaOH to give 0.5 mol. each of HCO₂H and MeOH. The sirupy compound is rather difficult to oxidize with HNO₃, has feeble add properties, forming salts (as with BaCO₃) which hydrolyze on evaporation of their solns. and gives by the Schotten-Baumann method a tribenzoyl derivative, needles from MeOH, m. 111°. With neutral carbonates (CaCO₃) the action is similar but less vigorous and there are indications of the formation of intermediate products. In dilute solns., NH₄OH acts on A like an alkali, H being evolved and the solution remaining clear; in concentrated

solns., however, a small amount of a curdy white precipitate is formed. In absolute

MeOH larger yields are obtained, but the best yields (up to 50% of the weight of A used) are secured in H₂O with NH₄Cl and a neutral carbonate or bicarbonate. The substance is a curdy, somewhat stringy, very voluminous precipitate, insol. in H₂O, dilute acids and alkalies and the ordinary organic solvents but on boiling with H₂O for several days it dissolves with formation of the NH₄ salt of a complex organic phosphoric acid; it is easily soluble in cold dilute HCHO, is quite stable to heat, suddenly swells far above 200°, gives off an odor of decayed fish and burns; it reacts violently with concentrated HNO₃; its composition varies somewhat (P

27-30, N

14.5-16.2%). The filtrate from this NH₃ derivative contains hexamethylenetetramine. PhNH₂ gives a similar precipitate with A.

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	111.56	111.77
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE	TOTAL

CA SUBSCRIBER PRICE	ENTRY	SESSION
	- 6.24	- 6.24

FILE 'USPATFULL' ENTERED AT 17:30:45 ON 21 AUG 2007
CA INDEXING COPYRIGHT (C) 2007 AMERICAN CHEMICAL SOCIETY (ACS)

FILE COVERS 1971 TO PATENT PUBLICATION DATE: 21 Aug 2007 (20070821/PD)
FILE LAST UPDATED: 21 Aug 2007 (20070821/ED)
HIGHEST GRANTED PATENT NUMBER: US7260849
HIGHEST APPLICATION PUBLICATION NUMBER: US2007192920
CA INDEXING IS CURRENT THROUGH 21 Aug 2007 (20070821/UPCA)
ISSUE CLASS FIELDS (/INCL) CURRENT THROUGH: 21 Aug 2007 (20070821/PD)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Jun 2007
USPTO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Jun 2007

=> s deodorant
 6740 DEODORANT
 7294 DEODORANTS
L28 11772 DEODORANT
 (DEODORANT OR DEODORANTS)

=> s phosphorus
 144499 PHOSPHORUS
 32 PHOSPHORUSES
 22 PHOSPHORI
 32 PHOSPHORIS
L29 144537 PHOSPHORUS
 (PHOSPHORUS OR PHOSPHORUSES OR PHOSPHORI OR PHOSPHORIS)

=> s odor
 51499 ODOR
 20725 ODORS
L30 61092 ODOR
 (ODOR OR ODORS)

=> s removal
 955394 REMOVAL
 3834 REMOVALS
L31 956265 REMOVAL
 (REMOVAL OR REMOVALS)

=> s (l29 (4w) 131 (4w) l29)
L32 46 (L29 (4W) L31 (4W) L29)

=> l28 and l32
L28 IS NOT A RECOGNIZED COMMAND
The previous command name entered was not recognized by the system.
For a list of commands available to you in the current file, enter
"HELP COMMANDS" at an arrow prompt (>).

=> s l28 and l32
L33 0 L28 AND L32

=> s acylaminoacid
 58 ACYLAminoACID
 30 ACYLAminoACIDS
L34 83 ACYLAminoACID
 (ACYLAminoACID OR ACYLAminoACIDS)

=> s 134 and 129
L35 4 L34 AND L29

=> s 135 and 130
L36 0 L35 AND L30

=> s 135 and 131
L37 1 L35 AND L31

=> d 137 IBIB

L37 ANSWER 1 OF 1 USPATFULL on STN
ACCESSION NUMBER: 2004:101834 USPATFULL
TITLE: Novel spla2 inhibitors
INVENTOR(S):
Bheight, Douglas Wade, Frankfort, IN, UNITED STATES
Kinnick, Michael Dean, Indianapolis, IN, UNITED STATES
Lin, Ho-Shen, Indianapolis, IN, UNITED STATES
Morin, John Michael, Brownsburg, IN, UNITED STATES
Richett, Michael Enrico, Indianapolis, IN, UNITED
STATES
Sall, Daniel Jon, Greenwood, IN, UNITED STATES
Sawyer, Jason Scott, Indianapolis, IN, UNITED STATES
Smith, Edward C R, Fishers, IN, UNITED STATES

	NUMBER	KIND	DATE
PATENT INFORMATION:	US 2004077704	A1	20040422
	US 6974831	B2	20051213
APPLICATION INFO.:	US 2003-450741	A1	20030616 (10)
	WO 2001-US43187		20011206

	NUMBER	DATE
PRIORITY INFORMATION:	US 2000-60256397	20001218
DOCUMENT TYPE:	Utility	
FILE SEGMENT:	APPLICATION	
LEGAL REPRESENTATIVE:	Francis O Ginah, Eli Lilly & Company, Patent Division, PO Box 6288, Indianapolis, IN, 46206-6288	
NUMBER OF CLAIMS:	26	
EXEMPLARY CLAIM:	1	
LINE COUNT:	2890	

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

=> d his

(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1 44 S ACYLAminoACID/IA
L2 5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?)/IA
L3 0 S L1 AND L2
L4 113543 S DETERGENT
L5 3 S L1 AND L4
L6 7262 S PHOSPHOROUS (2W) ACID
L7 1836 S HYPOPHOSPHOROUS (2W) ACID
L8 7262 S (L6 OR L7 (6W) ODOR)
L9 2 S L8 (5W) MEMBRANE

L10 73 S L8 AND L4
 L11 0 S L10 AND L1
 L12 0 S L10 (5W) ACYLAminoACID
 L13 7262 S (L6 OR L7 (2W) ODOR)
 L14 7273 S (L6 OR L7 AND ODOR)
 L15 7263 DUP REM L14 (10 DUPLICATES REMOVED)
 L16 113867 S GLUTAMIC
 L17 7263 S L15
 L18 23 S L15 AND L16
 L19 40 S FATTYACID
 L20 0 S L19 AND HALIDE
 L21 676258 S REMOVAL
 L22 7263 S L15
 L23 311 S L21 AND L15
 L24 2 S L23 (4W) L16
 L25 78807 S ODOR
 L26 7066 S L25 AND L21
 L27 4 S L26 AND L7

FILE 'USPATFULL' ENTERED AT 17:30:45 ON 21 AUG 2007

L28 11772 S DEODORANT
 L29 144537 S PHOSPHORUS
 L30 61092 S ODOR
 L31 956265 S REMOVAL
 L32 46 S (L29 (4W) L31 (4W) L29)
 L33 0 S L28 AND L32
 L34 83 S ACYLAminoACID
 L35 4 S L34 AND L29
 L36 0 S L35 AND L30
 L37 1 S L35 AND L31

=> file caplus

COST IN U.S. DOLLARS	SINCE FILE ENTRY	TOTAL SESSION
FULL ESTIMATED COST	20.11	131.88
DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)	SINCE FILE ENTRY	TOTAL SESSION
CA SUBSCRIBER PRICE	0.00	-6.24

FILE 'CAPLUS' ENTERED AT 17:38:19 ON 21 AUG 2007

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FILE COVERS 1907 - 21 Aug 2007 VOL 147 ISS 9
 FILE LAST UPDATED: 20 Aug 2007 (20070820/ED)

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<http://www.cas.org/infopolicy.html>

=> d 110 1-40 abs

- L10 ANSWER 1 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Bisphosphonates have found significant use in pharmaceutical industry as well as in ion sequestering and detergent applications. Blockbuster osteoporosis drug, sodium alendronate, is currently commercialized by several pharmaceutical companies. Cabot is interested in making aniline derivatized bisphosphonates for its applications. Herein we present scale-up process study of one particular mol., EBP, 2-(4-aminophenyl)-1-hydroxyethylidene-1,1-bisphosphonic acid, mono sodium salt. The synthesis of EBP involves the reaction of 4-aminophenylacetic acid (APAA) with phosphorous trichloride (PCl₃) and phosphorous acid in the presence of methane sulfonic acid (MSA). This is followed by hydrolysis and subsequent precipitation of EBP at a pH of around 4.3. A 12 factor, 2² experiment, screening design of expts. (DOE) was performed by varying the concns. of all the starting chems., addnl. water, temps., times at various stages of the reaction, and hydrolysis conditions. Results from screening DOE showed that increasing amts. of all the reactants relative to APAA, adding PCl₃ at a slower rate, reacting at higher temperature and reducing water of hydrolysis increased yield of EBP.

The reaction hold times could be reduced from 16 to 8 h without impacting yield. Yields could be increased from about 27% to 65% and results validated. Characterization and level reduction of major impurities from the reaction are also discussed.

- L10 ANSWER 2 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB It has been discovered that a lubricating oil composition containing a certain combination of a nitrogen-containing dispersant and a metal-containing detergent of an alkali metal salt of alkylphenol derivative having a Mannich base structure, together with a phosphorus-containing organic compound, can be effectively employed in a method for improving the acrylic rubber sealant compatibility in an internal combustion engine, if the ratio of the nitrogen-containing dispersant and a metal-containing detergent of an alkali metal salt of alkylphenol derivative having a Mannich base structure is adjusted to a specific range, i.e., in the range of from 1:0.005 to 1:2 in terms of the nitrogen content.

- L10 ANSWER 3 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB The present invention relates to a lubricating oil composition comprising a major amount of base oil of lubricating viscosity and the following additives in the following amts. based on a total amount of the lubricating oil composition: (A) 0.1 to 10 weight% of a succinic acid ester, (B) 0.01 to 2 weight% of a phosphorous acid ester, (C) 0.1 to 5 weight% of an amide compound which is a reaction product obtained by reaction of a linear or branched, saturated or unsatd. monovalent aliphatic acid having 8 to 22 carbon atoms, urea, and polyalkylenepolyamine, (D) 0.01 to 2 weight% of an overbased metal-containing detergent, and (E) 0.01 to 1 weight% of a corrosion inhibitor.

- L10 ANSWER 4 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Lubricant additives, especially friction modifiers, for automatic transmission fluids, are amides or thioamides of general structure R1R2N-C(:X)R3, in which X = O or S; R1 and R2 are C \geq 6-hydrocarbyl (preferably 2-ethylhexyl or C10-18-hydrocarbyl); and R3 = C1-6-hydroxyalkyl, prepared by an acylation reaction. The lubricant additives can also contain a nitrogen-containing dispersant, and a phosphorus-containing compound, provided that the phosphorus-containing compds. contain \leq 0.1 weight% zinc dialkyl dithiophosphates. Suitable phosphorus-containing compds. are phosphoric acid, phosphorous acid, phosphonic acid (or corresponding esters), dihydrocarbyl hydrogen phosphites, dihydrocarbyl dithiophosphate esters, trihydrocarbyl thiophosphates, or corresponding salts. The amide or thioamide additives are preferably present at the 0.2-5 weight% treating level. The lubricants can also contain a number of other additives (e.g., detergents, antioxidants, corrosion inhibitors, etc.).

L10 ANSWER 5 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB A lubricating composition contains: (1) a first base oil, derived from a gaseous source, with a viscosity index \geq 115, a sulfur content \leq 0.3 weight% S, and 95-100 weight% branched alkanes, (2) optionally a second base oil derived from a liquid petroleum source, (3) 1-30 weight% of a solubilizer selected from adipate esters, polyol esters, alkylated naphthalenes, alkylated sulfones, naphthenic base oils, aromatic base oils, and alkylated benzenes, and (4) an additive component. The base oil component contains 5-100 weight% of the first base oil. The additive component is selected from viscosity index improvers, dispersants, friction modifiers, corrosion inhibitors, rust inhibitors, antioxidants, detergents, seal swelling agents, extreme-pressure additives, antiwear additives, pour point depressants deodorizers, foam inhibitors, demulsifiers, dyes, thickeners, and fluorescent dyes.

L10 ANSWER 6 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Lubricating oils for the hydraulic system of a farm tractor consist of: (1) a base oil, (2) an amine salt of a phosphorus-containing acid ester, (3) a thiadiazole compound, (4) an overbased metal detergent, (5) a boron compound other than the overbased detergent, and (6) a friction modifier other than the boron compound, in which the entire lubricating oil is free from zinc dialkyl dithiophosphates. The amine salt is a C8-20-alkylamine salt of a mono- or dialkyl phosphate ester, typically prepared by reaction of a dialkyl dithiophosphate ester with P205. The thiadiazole is selected from 2-alkyldithio-5-mercapto-1,3,4-thiadiazoles, 2,5-bis(alkylthio)-1,3,4-thiadiazoles, and 2-alkyl(hydroxyphenylmethyl)thio-5-mercapto-1,3,4-thiadiazoles, optionally reacted with a nitrogen-containing dispersant. A suitable boron compound (component 5) is a borated ethoxylated amine or the reaction product of C8-20-fatty acids with dialkanolamines and boric acid.

L10 ANSWER 7 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The toothpaste is comprised of foaming agent 25-80, friction producing material 1-25, cleanser 0.01-20, bulking agent 1-30, and anti-dental calculus material 0.1-8 %. The foaming agent contains acidic substance 10-40 % which is selective from at least one of citric acid, tartaric acid and malic acid, and alkali substance 10-40 % which is selective from sodium bicarbonate. The cleanser is selective from at least one of sodium dodecyl sulfonate or potassium dodecyl sulfonate, poloxamer, steareth 30, and polysorbate 20. The friction producing material is selective from at least one of calcium carbonate, calcium phosphate, silicon dioxide,

aluminum hydroxide, and bentonite. The bulking agent is selective from at least one of xylitol, mannitol, and lactose. The anti-dental calculus material is selective from at least one of pyrophosphate, polyphosphoric acid salt, and cyclophosphoric acid salt. The toothpaste further contains 0.01-10 % germicidal agent from mint, clove, etc., anti-caries agent from sodium fluoride, potassium fluoride, etc., desensitizing agent from strontium chloride, and whitening agent from sodium borate, sodium percarbonate, etc. The process comprises mixing, pelleting to obtain granule, oven drying, and tabletting.

L10 ANSWER 8 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Cleaning and disinfectant compns. particularly for use with hard surfaces, include hydrogen peroxide and an acid or salt thereof which is resistant to oxidation other than phosphorous based acids.

Replacement of phosphorous based acids with acids of the invention results in improved hydrogen peroxide stability while maintaining or increasing the efficacy of cleaning and antimicrobial activity of the compns. Typically, acids of the invention include substituted or unsubstituted carboxylic acids such as R₄-C(R₂)(R₃)R₁-COOH, wherein R₁₋₄ are aroms.; R₁ is absent or is a substituted or unsubstituted alkylene, heteroalkylene, alkenylene, heteroalkenylene, alkynylene, or heteroalkynylene, each having up to 10 carbon atoms; R₂ and R₃ are each independently substituted or unsubstituted C₁₋₈ alkyl; and R₄ is substituted or unsubstituted alkyl, alkenyl, alkynyl, heteroalkyl, heteroalkenyl, or heteroalkynyl. Inventive compns. may further include surfactants, a chelating agent or sequestrant, a water soluble or water dispersible solvent, corrosion inhibitors and other adjuvants well known to those skilled in the art. There are further provided methods of use and methods of preparing inventive compns.

L10 ANSWER 9 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Power transmission fluids with improved anti-shudder properties contain an organic (thio)phosphate additive of general structure R₁-X₂-P(:X₁)(X₃-R₂)(X-R₅) (I), in which (1) R₁ and R₂ are C₁₋₂₄-alkyl, aryl alkylaryl or cycloalkyl that may also contain other atoms besides carbon and hydrogen (e.g., Cl, S, O, or N); (2) R₅ is derived from a reactive olefin, or is C-CH₂-CHR-C(:O)O-R₆, -CH₂-CH(R₇)(R₈), or R₉-O-C(:O)-CH₂-CH-C(:O)-O-R₁₀, where R is H or R₁, (3) R₆, R₇, R₉, and R₁₀ is the same as R₁; and (4) R₈ = Ph or C₆₋₃₀-alkyl or alkenyl-substituted Ph. The transmission fluid can also contain a calcium detergent, a friction modifier, and an organic phosphite of general structure R-O-PH(:O)(O-R₁) (R is hydrocarbyl, and R₁ = H or hydrocarbyl). I is preferably of structure (R-O)₂-P(:S)-S-CH(CO₂R₁)-CH₂-CO₂R₂, in which R, R₁, and R₂ = C₃₋₈-alkyl. The power transmission fluid can also contain such additives as bis(succinimide-terminated) polyethylenepolyamines, N-(substituted-alkyl)diethanolamines, N,O-disubstituted tetrahydro-2-hydroxy-4H-1,3,6,2-dioxazaborocines, and C₁₂₋₂₄-alkyl- and alkenylamides. The power transmission fluids are especially continuously variable transmission fluids.

L10 ANSWER 10 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB A detergent bar composition comprises: (a) from 15 to 70 % by weight of one or more detergent actives; (b) one or more hydrated phosphates; (c) from 0.1 to 5 % by weight of at least one water soluble salt of an organic carboxylic acid; and (d) from 10 to 55 % by weight of water; wherein the anhydrous component of hydrated phosphate is up to 20% phosphorus pentoxide (P₂O₅). A process for preparing a detergent bar composition comprises the steps of: (a) generating phosphate by reacting phosphorous containing mineral acid with an alkali in presence of the detergent active precursor at a temperature between 25

°C and 105 °C to obtain a mixture of phosphate and neutralized detergent active; (b) generating in or adding to the neutralized detergent active the water soluble salt of an organic carboxylic acid; (c) adding if desired, other detergent actives and minor additives to the detergent mixture; and (d) converting the product of step (c) into bars by conventional method.

L10 ANSWER 11 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Compds. comprising a 5-membered ring containing at least two double bonded nitrogen atoms can be used to impart improved metal pitting performance to a transmission fluid. The fluid includes an oil of lubricating viscosity and a hydrocarbyl ester of a phosphorous acid and is preferably limited to 0 to 0.1% by weight of zinc salts of sulfur-containing phosphorus acids.

L10 ANSWER 12 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB An acidic aqueous hydrogen peroxide solution is provided with improved disinfectant activity contain up to about 8 weight/weight % H₂O₂ and as-used concns. contain about 0.5 % weight/weight % H₂O₂. The solution also contains from

0.05 to 8.0 weight/weight % of at least one phosphorous-based acid, e.g. phosphoric acid and/or a phosphonic acid with from 1 to 5 phosphonic acid groups, and from 0.02 to 5 weight/weight % of at least one anionic surfactant. The surfactant is selected from C₈ to C₁₆ alkyl aryl sulfonic acids, sulfonated C₁₂ to C₂₂ carboxylic acids, C₈ to C₂₂ alkyl di-Ph oxide sulfonic acids, naphthalene sulfonic acids, C₈ to C₂₂ alkyl sulfonic acids, and alkali metal and ammonium salts thereof, and alkali metal C₈ to C₁₈ alkyl sulfates, and mixts. thereof. Most preferably the solution has an emulsifier and/or hydrotrope, e.g. an alkylated sulfonated di-Ph oxide salt, an alkyl aryl polyoxyethylene surfactant, and/or a polyoxyethylene surfactant. The solution may also contain corrosion inhibitors and/or lower alcs. A solution was prepared containing phosphoric acid,

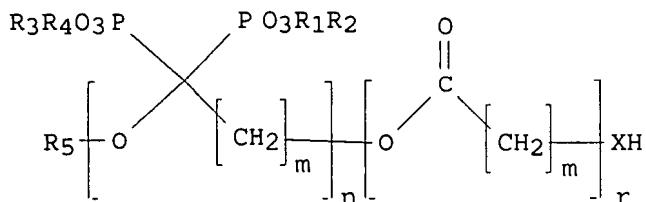
Briquest 301-50A [aminotri(methylenephosphonic acid)], H₂O₂, and surfactants.

L10 ANSWER 13 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The title composition is manufactured by blending a lubricating base oil with ≥1 Si compds. selected from (A) organic ortho silicate of formula: Si(OR₁₁)₄ (R₁₁ = C₂-30 hydrocarbyl); (B) organic ortho silica-polyamine condensate of formula: (R₂₁₀)_xSi[(NH₂CpH₂p)_qNHR₂₂]_{4-x} (R₂₁ and R₂₂ = C₁-30 hydrocarbyl; p = an integer of 1-36; q = an integer of 1-4; x = an integer of 0-3), and (C) organic ortho silicate-polyol condensate of formula: (R₃₁₀)_ySi[(OR₃₂(OH)_r)SOH]_{4-y} (R₃₁ = C₁-30 hydrocarbyl; R₃₂ = C₂-36 hydrocarbyl; r = an integer of 0-2; s = an integer of 1-4; y = an integer of 0-3), and optionally (D) an ashless dispersant such as alkenylsuccinimides; (E) metal soap-type detergent and/or (F) extreme-pressure additive such as dialkyl dithiophosphates; (G) antiwear additive such as phosphate esters; (H) antifriction additive, (I) antioxidant, and (J) viscosity-index improver such as polymethacrylates. The composition is superior in achieving high transmission torque capacity, durability, and antifriction-antiwear performance.

L10 ANSWER 14 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

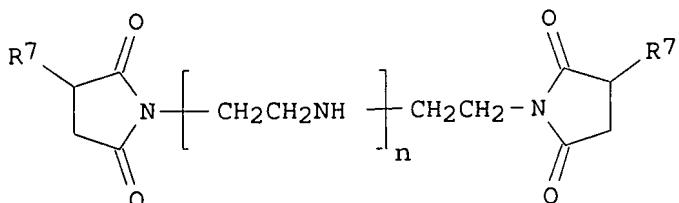
GI



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AB The preparation of geminal-diphosphonic acids [I; wherein R1, R2, R3, and R4, independently = H, Na, or K; R5 = H or (C1-C10)alkylidene comprising at least one hydroxyl substituent; m = 3, 4, 5, or 6; n ≥ 1; r ≥ 0; X = O, PO4H, or PO3H] is described. Thus, ε-caprolactone and phosphorous acid were reacted in the presence of phosphorous trichloride, with subsequent treatment with NaOH, to give oxocycloheptane-2,2-diphosphonic acid, sodium salt. The compds. are claimed to be useful as sequestrants, corrosion inhibitors, in detergent compns., treatment of certain diseases caused by the imbalance of metal cations in the body, cleaning compns., cosmetic and toiletries, boiler water treatment, cooling water treatment, desalination, and oil recovery. The prepared compds. showed calcium sequestration results at 20 g calcium/100 g sequestrant of up to %97.5.

L10 ANSWER 15 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
GI



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AB A zinc-free lubricating composition for a continuously variable vehicle transmission contains: (1) a base oil, (2) a ashless polyisobutylene succinimide dispersant (with mol. weight 700-1200), (3) an organic ether phosphite, (4) an overbased calcium phenate detergent (at Ca concentration of <500 ppm), (5) succinimide and a N,N-bis(2-hydroxyethyl) amine friction modifiers, and (6) a long-chain primary carboxamide. The ether phosphite has a general structure R-O-PH(=O)-OR1 (R is hydrocarbyl, R1 = H or hydrocarbyl, and R or R1 contains a thioether group). Succinimides of component (5) have the structure I (R7 = C6-30-alkyl; z = 1-10), and the ethoxylated amines have the structure R8-X-(CH2)x-N(CH2CH2OH)2 (R8 = C6-28-alkyl; X = O, S, or CH2; x = 1-6).

L10 ANSWER 16 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB An aqueous hard surface cleaning and disinfecting composition comprises: a first acid sequestrant constituent which comprises citric acid and a second acid sequestrant which is selected from citric acid, cresylic acid,

dodecylbenzene sulfonic acid, phosphoric acid, salicylic acid, sorbic acid, sulfamic acid, acetic acid, benzoic acid, boric acid, capric acid, caproic acid, cyanuric acid, dihydroacetic acid, dimethylsulfamic acid, propionic acid, polyacrylic acid, 2-Et hexanoic acid, formic acid, fumaric acid, 1-glutamic acid, iso-Pr sulfamic acid, naphthenic acid, oxalic acid, phosphorus acid, valeric acid, benzene sulfonic acid, xylene sulfonic acid, sulfonic acids, maleic acid, acetic acid, adipic acid, lactic acid, butyric acid, gluconic acid, malic acid, tartaric acid, and glycolic acid; a mixture of hydrophobic and hydrophilic solvents wherein the hydrophobic solvent exhibits a solubility in water of from 0.0 - 20.0 mL/100mL of water, and which comprises 51-99% of the mixture of solvents, and wherein the hydrophilic solvent comprises 1-49% of the mixture of solvents; 0.001-1% by weight of a single constituent which exhibits both anionic surfactant and hydrotrope properties; 0-20% of one or more optional constituents; the balance to 100% by weight, water; wherein the composition has pH ≤7.0. The compns. provide good removal of soap scum stains, and further feature low levels of irritability to the user. Although acidic, the improved hard surface cleaning compns. feature low irritability to the eyes and skin of consumers. The compns. also provide disinfecting effects. Processes for the production of the compns., as well as methods for their use are also described.

L10 ANSWER 17 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The invention concerns a method for obtaining homopolymers and/or copolymers in aqueous solution using phosphorous acid and/or its salts or sodium hypophosphite and by neutralizing during polymerization ethylenically unsatd. acid monomers by continuously adding firstly bases such as sodium hydroxide, potassium hydroxide or lithium hydroxide, then alkaline-earth bases such as calcium hydroxide, magnesium hydroxide, calcium oxide or magnesium oxide. The mol. weight of the polymers is a function of the concentration of the initiators and the P-containing compds., and the polymers are useful as dispersants for inorg. compds. in aqueous media.

L10 ANSWER 18 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The cleaning compns. (A) contain H₃PO₄ and/or phosphoric acid salts and have pH 0.1-6.0 or the cleaning compns. comprise A compns. containing chelating agents or the cleaning compns. comprise A compns. containing organic acids. A silicon wafer was stained on the surface with a solution containing 20 ppm Fe and 20 ppm Cu, immersed in aqueous composition (pH 3) containing H₃PO₄ 5, 1-hydroxydiethylidene-1,1-diphosphonic acid 3, and oxalic acid 3%, and treated with ultrasonic waves for 5 min at 38 kHz and washed for 2 cycles to give a cleaned wafer with residual metal impurity content <10⁹ atoms/cm².

L10 ANSWER 19 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Title compns. are reduced in hygroscopicity by mixing R₁CHR₂NR₃CHR₄CO₂M (R₁, R₃ = H, Me, CH₂OH, CO₂M, CH₂CO₂M, CH(OH)CO₂M, CHMeCO₂M, etc.; R₂ = CO₂M, CH₂CO₂M, CH(OH)CO₂M, CHMeCO₂M, etc.; R₄ = H, Me, OH, CH₂OH; M = H, NH₄, alkali metal salt) with H₃BO₃, H₂CO₃, HNO₃, H₂SO₄, H₂SO₃, HCl, silicic acid, H₃PO₄, H₂PHO₃, and/or carboxylic acids to adjust pH ≤10 and drying. Thus, 1 kg aqueous solution of 40% L-aspartic acid-N,N-diacetic acid tetrasodium salt was mixed with 119 g H₂SO₄ and dried to give a composition showing hygroscopicity 6.6% after 14 h at 25° and humidity 55%.

L10 ANSWER 20 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB Surface regeneration of affinity biosensors and characterization of biomols. associated therewith by multivariate technique employing cocktails of regeneration agents to optimize regeneration of biosensor surface and/or characterize biomols. associated therewith. Kits and stock solns. for use in the context of this invention, as well as associated computer algorithms are also disclosed. Stock solns. of regeneration cocktails are prepared and combined. Solns. are acidic, basic, ionic, organic, detergent and chelating agent containing Biosensors for various affinity bindings are regenerated by the method; the affinity reactions are used for optimizing the regeneration process. Immuno-reactions, nucleic acid hybridization, avidin/streptavidin-biotin, hormone-hormone receptor interactions are performed with Biocore instruments and CM5 sensor chips.

L10 ANSWER 21 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB An enzymic detergent drain cleaner composition or cleaner for aqueous systems comprises .apprx.0.005-20% acid lipase having hydrolytic activity to ester bonds, specific to glyceride linkages, .apprx.0.005-20% acid cellulase enzyme having hydrolytic activity specific to β -glucosidic bonds, .apprx.1-70% water-soluble carbonate salt, .apprx.1-70% water-soluble acid that reacts in an aqueous medium with the carbonate salt to form CO₂ that dissolves in the aqueous medium, .apprx.0.1-10% surfactant, and .apprx.0.05-5% thickening agent. A typical cleaner contained citric acid 39, Na₂CO₃ 5, NaHCO₃ 39, Nacconol 90G 10, Carbopol EZ-2 1, cellulase TR 4, and lipase BCC 2%.

L10 ANSWER 22 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The interaction between natural flake graphite and oil soluble additives and the effect of the interaction on the lubrication property of oils were studied. The results showed that the oil containing graphite only exhibits poor lubricity, but the oil containing both graphite and additive with active elements or group gives much better lubricity. This indicated that there exist good compatibility and some synergism between graphite and oil additives. The adsorption content of S element on graphite from the mixture of graphite and additives containing S was detected by XPS anal. It was found that the adsorption of organic sulfides on graphite was affected by the mol. structure of sulfides and the adsorption was beneficial for the lubricity. This could be attributed to the formation of the composite film of graphite and additives on the rubbing surfaces.

L10 ANSWER 23 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The cocondensates, useful as builders in low-P or phosphate-free detergents, are prepared by introducing aspartic acid into an extruder, injecting the amine and an acid catalyst, and mixing at $\leq 300^\circ$ such that the conversion of the aspartic acid within the extruder is 20-95%, followed by further condensation at 150-300° after extrusion. Thus, L-aspartic acid as particles of diameter .apprx.0.5 mm was continuously fed to an extruder, to which were sep. added tallow amines and H₃PO₄ at 80°, the temperature rising to 270° by the exit, and the extruded precondensate was condensed for 1 h at 180° to give 100% conversion of aspartic acid to polymer with weight-average mol. weight .apprx.8000.

L10 ANSWER 24 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

AB The title enzymic detergent drain cleaner composition comprises: 0.015-20% of an acid cellulase enzyme having hydrolytic activity specific to β -glucosidic bonds, 1-70% of a water soluble carbonate salt, 1-70% of a water soluble acid that reacts in an aqueous medium with the carbonate salt

to

form carbon dioxide that dissolves in the aqueous medium, 0.1-10% of a surfactant, and 0.05-5% of a thickening agent. This detergent composition may be used as an enzymic detergent drain cleaner or in a method for removing or preventing bacterial cellulose deposits in an aqueous system at a solution temperature of up to about 60° and a pH 2-7.

- L10 ANSWER 25 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB The process comprises polycondensing a monomer selected from aspartic acid, maleamic acid, a reaction product of maleic acid and ammonia, and mixts. in the presence of a specific phosphorous acid compound, preferably tri-Ph phosphite or tri-Bu phosphite. Polysuccinimide having a reduced residual monomer content and an increased mol. weight is produced efficiently. This compound can be used as a chelating agent, a scale inhibitor, a builder, a dispersant, a humectant or an additive for fertilizers.
- L10 ANSWER 26 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Aspartic acid is (co)condensed in an inert diluent in the presence of an acid catalyst at $\geq 120^\circ$ with removal of the water formed, followed by hydrolysis of the imide units formed. The (co)condensates are used as additives to cleaning compns., dispersants for detergent compns., and scale inhibitors during desalination of seawater. Thus, addition of 40 g tallow fatty amines and 212 g 75% H₃PO₄ sep. to a mixture of 600 g paraffin oil and 400 g aspartic acid at 170° and heating for 2 h until no more H₂O evolved gave a polycondensate containing 1.9% bound P, which was hydrolyzed with NaOH at 60° to give an aqueous solution (32% solids) of a hydrolyzed polycondensate with weight-average mol. weight 9000.
- L10 ANSWER 27 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB The compns., having minimal irritability to skin or eyes and pH ≤ 5.0 , comprise 0.1-10% an acid sequestrant; 0.1-10% mixture of hydrophobic and hydrophilic solvents; 1-8% surfactant and/or hydrotrope; 0-20% ≥ 1 optional constituents; the balance 100% H₂O. A cleaner containing Polytergent SL 62 1.00, Rhodapon LCP 3.00, Polytergent 2A1 3.00, Dowanol PnP 0.90, Dowanol PnB 3.90, citric acid 2.50, glycolic acid 3.57, fragrance 0.20, and water 81.93 had 63% cleaning efficiency (calculated from reflectance values); vs. 2.1% for a cleaner without glycolic acid or citric acid.
- L10 ANSWER 28 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB The materials are composed of (A) sulfosalicylic acid (I) and/or its alkali, NH₄, or lower amine salts, or (B) phosphorous acid mono- or di- lower or middle alkyl esters or their alkali, NH₄, or lower amine salts, or (C) lower or middle tetraalkylammonium salts of halogens, inorg. compds., lower carboxylic acids, lower alkylsulfonic acids, and lower alkylsulfuric acids. Aqueous compns. of surfactants, builders, etc., obtained with the solidifying agents are also claimed. Thus, an aqueous composition containing I monoethanolamine salt 5.0, Na alkylbenzenesulfonate 2.5, and Na₂CO₃ 3.0% showed good transparency at 5-25°.
- L10 ANSWER 29 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Aspartic acid is polymerized at $\geq 140^\circ$ in the presence of phosphorous or hypophosphorous acid to give a polymer which shows good biodegradability and is useful in laundry detergent compns. for improving the removal of clay soils from fabrics.
- L10 ANSWER 30 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN

- AB He title polymers, containing the groups $-\text{CH}_2\text{C}(\text{R}_2)[\text{CON}(\text{R}_1)\text{CH}_2\text{PO}(\text{OX})_2]-$ [$\text{R}_1 = \text{H}$, alkyl, $\text{CH}_2\text{PO}(\text{OX})_2$; $\text{R}_2 = \text{H}$, alkyl; $\text{X} = \text{H}$, NH_4 , alkali metal, alkaline earth], are prepared from the corresponding alkenamide polymers, HCHO , and H_3PO_3 . The reaction of polyacrylamide (K value 29.4), H_3PO_3 , and HCHO in mol ratio 1:2:8 at 80-100° gave a polymer containing 4.10% P. Use of the polymers as encrustation inhibitors in detergents and as scale inhibitors in water treatment is exemplified.
- L10 ANSWER 31 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
- AB The title compns. showing good discoloration resistance during storage contain amidoamino acid(s) chosen from $\text{R}_1\text{CONR}_2\text{CH}_2\text{CH}_2\text{NR}_3\text{CH}_2\text{CH}_2\text{CO}_2\text{M}$ and $\text{R}_1\text{CONR}_2\text{CH}_2\text{CH}_2\text{N}(\text{CH}_2\text{CH}_2\text{CO}_2\text{M})_2$ [$\text{R}_1 = \text{C}_7\text{-21}$ (un)saturated aliphatic hydrocarbyl, R_2 , $\text{R}_3 = \text{H}$, C1-4 hydroxyalkyl; M = H, Na, K, triethanolamine residue], optionally poly(acrylic acid) or its water-soluble salts, and phosphite(s) $\text{R}_4\text{O}(\text{R}_5\text{O})(\text{R}_6\text{O})\text{P}$ ($\text{R}_4\text{-6} = \text{H}$, Na, K, NH_4). A dishwashing liquid detergent comprised N-cocoalkyl-N'-(2-hydroxyethyl)-N'-(sodiocarboxyethyl)ethylenediamine 50, phosphorous acid 1.0, K phosphite 0.2, Na α-olefinsulfonate 5, laurylamine oxide, methylparaben 0.2, and water to 100%.
- L10 ANSWER 32 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
- AB Polymers useful as scale inhibitors and detergent builders are prepared from the phosphonic acids $\text{R}_4\text{R}_5\text{C}(\text{R}_6)\text{CH}(\text{R}_3)\text{NHCH}(\text{R}_1)\text{PO}_3\text{M}_2$ ($\text{R}_1 = \text{H}$, alkyl, aryl, hydroxyaryl; R3, R4, R6 = H, Me; R5 = H, alkyl, Ph; M = H, cation) and unsatd. carboxylic acids or their salts. Adding 72.5 g $\text{PhCH}:\text{NCH}_2\text{CH}:\text{CH}_2$ over 20 min to H_3PO_3 82, AcOH 100, and Ac₂O 20 g at 100° and refluxing for 3 h gave 82-87 g $\text{CH}_2:\text{CHCH}_2\text{NHCH}(\text{Ph})\text{PO}_3\text{H}_2$ (I). K₂S₂O₈-initiated polymerization of a solution of I 0.36, acrylic acid 36, 20% NaOH 100, and H₂O 125 g at 80° gave a 17% solution of copolymer (K-value 40-45), a solution of 1, 3, 10, and 30 ppm of which dispersed 26, 41, 61, and 90%, resp., of CaCO₃.
- L10 ANSWER 33 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
- AB The title polymers, having repeat units $\text{CH}_2\text{CHN}(\text{R}_1)\text{CH}_2\text{P}(:\text{O})(\text{OX})_2$ [$\text{R}_1 = \text{H}$, C1-6 alkyl, $\text{CH}_2\text{P}(:\text{O})(\text{OX})_2$; X = H, alkali metal, ammonium, alkaline earth metal], are prepared and are useful as water-treatment chems., laundry bleach stabilizers, and detergent formulation chems. Thus, 500 g of isopropanol was heated to boiling, and, over 3 h, a solution of 270.4 g acrylic acid and 29.6 g N-vinylformamide dissolved in 100 g isopropanol were added along with a solution of 9 g tert-Bu perethylhexanoate in 100 g isopropanol, the isopropanol azeotropically distilled off, producing a 27% aqueous polymer solution, 125 parts of which was reacted with 150 parts concentrated HCl at reflux and esterified with phosphorous acid and neutralized with NaOH, producing a solid product which had K value (Na salt, 1% in H₂O) 25.
- L10 ANSWER 34 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
- AB Lubricating oil and grease compns. having improved antiwear and extreme-pressure properties comprise salts prepared from a compound of the formula $\text{RX}(\text{CH}_2)_n\text{C}(\text{OH})\text{Y}\text{P}(:\text{O})(\text{OH})_2$, where X is D, S, or N in which the N may contain a H or alkyl substituent, n = 1-8 integer, Y is H, alkyl or a phosphonic acid group, and R is a C1-100 alkyl group, and a base selected from (A) a detergent, (B) a dispersant, which is chosen from (a) a Mannich, (b) a succinimide, (c) a N-containing ester dispersant, and (d) a dispersant-viscosity improver, and (C) an amine.

- L10 ANSWER 35 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Water-soluble polymers prepared by polymerizing 100 parts unsatd. monomers in the presence of 6-60 parts hypophosphorous and/or phosphorous acid or salt are useful as detergent builders. Adding 80% aqueous acrylic acid solution 1250, 3% aqueous $(\text{NH}_4)_2\text{S}_2\text{O}_8$ solution 422, and 30% aqueous Na hypophosphite (I) solution 267 parts to 3062 parts water at 70% gave a polymer (mol. weight 25,000). A detergent comprising the polymer 12, Na alkylbenzenesulfonate 30, Na silicate 10, Na_2CO_3 5, CM-cellulose 1, and Na_2SO_4 61 parts showed detergency (in washing soiled polyester-cotton blend) 73%, vs. 68 with a polymer prepared without I.
- L10 ANSWER 36 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB A catalyst system comprising >1 P-containing acid and >1 Al alcoholate or phenolate is useful for the alkoxylation of C6-30 alkanols by C2-4 alkylene oxides, giving alkoxylates (especially ethoxylates) which have a narrow mol. weight distribution of alkylene oxide adducts and a low content of unreacted alkanol and are useful as surfactants, especially in detergent comps. H_3PO_4 and (iso-PrO)₃Al were used as catalysts for the ethoxylation of 0.773 mol C12-13 alkanols by 2.2 mol ethylene oxide, giving ethoxylated alcs. having alkanol content 4.4%, polyethylene content 0.6%, and a narrow ethoxylate distribution.
- L10 ANSWER 37 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB A review with 54 refs. on H_3PO_3 dyes, sequestering agents, antistatic agents, detergent builders, dyeing buffers, H_2O_2 stabilizers, crosslinking catalysts, and P(III) anal.
- L10 ANSWER 38 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Reactors used for manufacture of ethylene terephthalate-based polyesters are cleaned by heating under pressure in the presence of ethylene glycol (I) [107-21-1] and H_3PO_3 . Thus, after 100 batches of poly(ethylene terephthalate) (II) [25038-59-9] manufacture, a 3000-L (capacity) reactor was cleaned with 2500 L I and 50 kg H_3PO_3 at 240° and 2.55 kg/cm² for 5 h to remove the black deposits. II prepared after cleaning contained no black particles which were observed frequently in II prepared before cleaning.
- L10 ANSWER 39 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Reactors used for manufacture of ethylene terephthalate-based polyesters are cleaned with mixts. of ethylene glycol (I) [107-21-1] and H_3PO_3 . Thus, after 100 batches of poly(ethylene terephthalate) (II) [25038-59-9] manufacture, a 3000-L reactor was cleaned with 2500 L I and 30 kg H_3PO_3 at 197° for 6 h to remove the black deposits. II prepared after cleaning contained no black particles (in a 30 + 30 + 0.3-cm sheet), compared with 10 particles ($\geq 100 \mu\text{-diameter}$) in II prepared before cleaning.
- L10 ANSWER 40 OF 73 CAPLUS COPYRIGHT 2007 ACS on STN
AB Detergents R(OCH₂CH₂)_nO₂CR₁ (R = saturated or unsatd., normal or branched C₄-30 alkyl or phenyl substituted with saturated or unsatd., normal or branched C₁-18 alkyl; R₁ = saturated or unsatd., normal or branched C₁-30 alkyl; n = 1-30) are prepared by esterification of R(OCH₂CH₂)_nOH with R₁CO₂H at 100-250° in an inert atmospheric, preferably in the presence of H_3PO_3 catalyst. Thus, Rokanol L-10 [9002-92-0] 100, stearic acid [57-11-4] 45.5, and H_3PO_3 1.45 g were heated at 200° in N to give a light yellow, waxy product [34383-57-8] having esterification degree 96%, and the product was soluble in water and lowered its surface tension.

=> d his

(FILE 'HOME' ENTERED AT 16:48:13 ON 21 AUG 2007)

FILE 'CAPLUS' ENTERED AT 16:48:58 ON 21 AUG 2007

L1 44 S ACYLAminoACID/IA
L2 5 S (FRAGRANCE (2W) COSMETIC (2W) TOILETR?) /IA
L3 0 S L1 AND L2
L4 113543 S DETERGENT
L5 3 S L1 AND L4
L6 7262 S PHOSPHOROUS (2W) ACID
L7 1836 S HYPOPHOSPHOROUS (2W) ACID
L8 7262 S (L6 OR L7 (6W) ODOR)
L9 2 S L8 (5W) MEMBRANE
L10 73 S L8 AND L4
L11 0 S L10 AND L1
L12 0 S L10 (5W) ACYLAminoACID
L13 7262 S (L6 OR L7 (2W) ODOR)
L14 7273 S (L6 OR L7 AND ODOR)
L15 7263 DUP REM L14 (10 DUPLICATES REMOVED)
L16 113867 S GLUTAMIC
L17 7263 S L15
L18 23 S L15 AND L16
L19 40 S FATTYACID
L20 0 S L19 AND HALIDE
L21 676258 S REMOVAL
L22 7263 S L15
L23 311 S L21 AND L15
L24 2 S L23 (4W) L16
L25 78807 S ODOR
L26 7066 S L25 AND L21
L27 4 S L26 AND L7

FILE 'USPATFULL' ENTERED AT 17:30:45 ON 21 AUG 2007

L28 11772 S DEODORANT
L29 144537 S PHOSPHORUS
L30 61092 S ODOR
L31 956265 S REMOVAL
L32 46 S (L29 (4W) L31 (4W) L29)
L33 0 S L28 AND L32
L34 83 S ACYLAminoACID
L35 4 S L34 AND L29
L36 0 S L35 AND L30
L37 1 S L35 AND L31

FILE 'CAPLUS' ENTERED AT 17:38:19 ON 21 AUG 2007

=> s 126 and 16
L38 8 L26 AND L6

=> d 138 1-8 IBIB

L38 ANSWER 1 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
ACCESSION NUMBER: 1992:553040 CAPLUS
DOCUMENT NUMBER: 117:153040
TITLE: Chemically modified cellulosic compositions for molding

INVENTOR(S) : Takechi, Shigetoshi; Mizomoto, Michihisa; Ueda, Minoru; Matsuda, Hideaki
 PATENT ASSIGNEE(S) : Okura Industrial Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 8 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 04122744	A	19920423	JP 1990-242526	19900914
PRIORITY APPLN. INFO.:			JP 1990-242526	19900914
OTHER SOURCE(S) :	MARPAT 117:153040			

L38 ANSWER 2 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1990:617267 CAPLUS
 DOCUMENT NUMBER: 113:217267
 TITLE: Deodorant and method for removing odors from tank trucks for collection of night soil by the vacuum method
 INVENTOR(S) : Yoshida, Norikazu; Matsumoto, Susumu
 PATENT ASSIGNEE(S) : Nippon Kayaku Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 6 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 02160019	A	19900620	JP 1988-312845	19881213
PRIORITY APPLN. INFO.:			JP 1988-312845	19881213

L38 ANSWER 3 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1990:164277 CAPLUS
 DOCUMENT NUMBER: 112:164277
 TITLE: Deodorant compositions with microbicidal activities capable of removing mercaptan odor
 INVENTOR(S) : Miki, Yoshiaki; Ueda, Tsunehisa; Natsume, Yoshio
 PATENT ASSIGNEE(S) : Nippon Zeon Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 4 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01124460	A	19890517	JP 1987-281372	19871107
PRIORITY APPLN. INFO.:			JP 1987-281372	19871107

L38 ANSWER 4 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1990:164276 CAPLUS
 DOCUMENT NUMBER: 112:164276
 TITLE: Deodorant compositions capable of removing

INVENTOR(S): odors of basic materials and mercaptans
 Miki, Yoshiaki; Ueda, Tsunehisa; Natsume, Yoshio
 PATENT ASSIGNEE(S): Nippon Zeon Co., Ltd., Japan
 SOURCE: Jpn. Kokai Tokkyo Koho, 3 pp.
 CODEN: JKXXAF
 DOCUMENT TYPE: Patent
 LANGUAGE: Japanese
 FAMILY ACC. NUM. COUNT: 1
 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 01124459	A	19890517	JP 1987-281371	19871107
JP 2549876	B2	19961030		
PRIORITY APPLN. INFO.:			JP 1987-281371	19871107

L38 ANSWER 5 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1961:130689 CAPLUS
 DOCUMENT NUMBER: 55:130689
 ORIGINAL REFERENCE NO.: 55:24531i,24532a-b
 TITLE: Hydrolysis of trialkyl phosphites in the presence of inorganic and organic bases
 AUTHOR(S): Imaev, M. G.
 CORPORATE SOURCE: S. M. Kirov Chem.-Technol. Inst., Kazan
 SOURCE: Zhurnal Obshchey Khimii (1961), 31, 1767-70
 CODEN: ZOKHA4; ISSN: 0044-460X
 DOCUMENT TYPE: Journal
 LANGUAGE: Unavailable

L38 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1960:110273 CAPLUS
 DOCUMENT NUMBER: 54:110273
 ORIGINAL REFERENCE NO.: 54:20966i,20967a-i,20968a-b
 TITLE: The decomposition of diacyl hydroxylamines and hydrazines
 AUTHOR(S): Walling, Cheves; Naglieri, Antonny N.
 CORPORATE SOURCE: Columbia Univ.
 SOURCE: Journal of the American Chemical Society (1960), 82, 1820-5
 CODEN: JACSAT; ISSN: 0002-7863
 DOCUMENT TYPE: Journal
 LANGUAGE: Unavailable

L38 ANSWER 7 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1953:61776 CAPLUS
 DOCUMENT NUMBER: 47:61776
 ORIGINAL REFERENCE NO.: 47:10461c-i,10462a-e
 TITLE: Substituted amides of cyclic esters of phosphorous acid
 AUTHOR(S): Arbuzov, A. E.; Zoroastrova, V. M.
 CORPORATE SOURCE: Kazan State Univ.
 SOURCE: Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya (1952) 789-800
 CODEN: IASKA6; ISSN: 0002-3353
 DOCUMENT TYPE: Journal
 LANGUAGE: Unavailable

L38 ANSWER 8 OF 8 CAPLUS COPYRIGHT 2007 ACS on STN
 ACCESSION NUMBER: 1953:698 CAPLUS

DOCUMENT NUMBER: 47:698
ORIGINAL REFERENCE NO.: 47:98b-g
TITLE: Preparation and properties of esters of
[bicyclohexyl]-1,1'-diol phosphorous
acid
AUTHOR(S): Arbuzov, A. E.; Azanovskaya, M. M.
CORPORATE SOURCE: Kazan State Univ.
SOURCE: Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya
(1951) 544-50
CODEN: IASKA6; ISSN: 0002-3353
DOCUMENT TYPE: Journal
LANGUAGE: Unavailable

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CA SUBSCRIBER PRICE	-31.20	-37.44

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